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The effect of powdering on the polytypic crystal structures of tin disulphide

During a study of polytypism in single crystals of tin disulphide grown by the iodine vapour transport method, single crystal X-ray diffraction studies brought to light several instances of polytypic structure, and in one or two cases revealed hitherto unreported polytypes. A full account of this work is to appear elsewhere [1]. A parallel study of similar samples by means of powder X-ray diffraction failed to reveal any indication of polytypes other than the 2H form [2].

In view of this, it was thought that the process of crushing and grinding in an agate mortar, adopted for the preparation of powder specimens, might be affecting the atomic structure of the resulting samples, possibly by the introduction of numerous stacking faults and dislocations to alter the structure of the higher order polytypes. The structure of tin disulphide is isomorphous with cadmium iodide, and is characterized by easy cleavage and glide parallel to the basal plane of the trigonal crystals.

To test this hypothesis, three SnS₂ crystals were carefully examined by single crystal diffraction methods to discover their polytype structures, and they were then crushed to make powder specimens. The powder diffraction data were then compared with those from the single crystals to determine whether the same polytypes were still present.

The powder data derived from a crystal known from single crystal studies to consist entirely of the 2H polytype (P $\bar{3}$ m1; Zhdanov symbol [11], $a = b = 3.643 \pm 0.002$ Å, $c = 5.894 \pm 0.005$ Å) were in excellent agreement with those calculated for the 2H structure. The comparison is made in Table I. Absorption and temperature factor cor-

rections have not been applied, and the relative intensities only of diffraction lines which are fairly close together may be quantitatively compared. The structure of this crystal is therefore unaffected by the powdering process.

TABLE I A comparison of the experimentally observed powder data for SnS₂ crystal no. 1 (2H polytype) with values calculated on the basis of the 2H structure (CuK α radiation)

Reflection <i>hkl</i>	θ_{obs} (deg)	θ_{calc} (deg)	d_{calc} (Å)	I_{calc} (normal- ized)	I_{obs}
001	7.48	7.51	5.894	94.7	vvs
100	14.10	14.14	3.155	31.6	ms
002	15.05	15.16	2.947	3.1	vw
{ 101 10 $\bar{1}$	16.07	16.09	2.782	11.5 100.0 }	vvs
102	20.93	20.97	2.154	69.5	ms
003	23.10	23.10	1.965	5.4	vw
110	24.98	25.04	1.822	29.8	m
111	26.25	26.29	1.740	22.0	mw
{ 103 10 $\bar{3}$	27.46	27.53	1.668	20.2 2.8 }	mw
200	29.17	29.25	1.577	4.3	vw
112	29.85	29.84	1.549	3.4	vw
{ 201 20 $\bar{1}$	30.32	30.30	1.524	14.9 2.1 }	mw
004	31.42	31.55	1.473	4.8	vw
202	33.54	33.66	1.391	16.4	mw
{ 113 104	35.21	35.25 35.27	1.336 1.335	9.0 4.9 }	w
{ 203 20 $\bar{3}$	38.75	38.81	1.230	1.0 7.2 }	vw-w
210	40.16	40.28	1.192	3.3	vw
005	40.68	40.84	1.179	1.0	vw
{ 121 211	41.16	41.27	1.169	12.4 1.7 }	mw
114	42.21	42.29	1.146	12.9	w
{ 212 105 10 $\bar{5}$	44.09	44.22 44.28 44.28	1.105 1.104 1.104	16.0 0.7 5.3 }	m

A crystal consisting almost entirely of 4H prior to crushing was also examined. The crystal contained a very small region of 40H polytype, of undetermined stacking sequence. Powder data obtained from this crystal are compared in Table II with those calculated on the basis of the 4H structure (P₆mc, Zhdanov symbol [22]; $a = b = 3.643 \pm 0.002 \text{ \AA}$, $c = 11.79 \pm 0.01 \text{ \AA}$).

The level of agreement between the two sets of data is poor. In particular, the observed intensity of the 10.0 reflection is very high compared with the corresponding calculated value, whilst the 10.2 and 10.5 reflections, although theoretically expected to be of fairly high intensity, could not be detected experimentally.

The possibility that the intensity discrepancies might be explicable by preferred orientation in the sample is discounted. To fabricate a powder

specimen in which preferred orientation did not occur the crystal was powdered. The powder was then mixed with Canada balsam and allowed to dry. The dried sample was then re-powdered before rolling the equi-axed granular material on to a fine glass fibre. No arcing characteristic of preferred orientation was visible on the powder photographs.

The influence of possible polytype transformations induced by the powdering process was therefore considered. A comparison of Tables I and II shows that the strong 10.0 reflection of the 2H polytype overlaps the weaker 10.0 reflection of the 4H polytype, and could give rise to an increase in the observed intensity. However, other strong 2H reflections, such as 10.1 and 10.1̄ are not observed. Intensity comparisons were calculated for a number of differing polytypes and polytype mixtures, but without acceptable agreement.

The results obtained, therefore, show that the structure of the 4H polytype was indeed influenced by the crushing process, but in a way which cannot be explained simply in terms of the formation of alternative polytypes.

A crystal containing the 4H, 6H and 18R polytypes prior to crushing was examined. The data relating to each type are as follows:

4H P₆mc, Zhdanov symbol [22]; $a = b = 3.643 \pm 0.002 \text{ \AA}$, $c = 11.79 \pm 0.01 \text{ \AA}$.

6H (Stacking sequence not positively identified). Either P₃m̄l, Zhdanov symbol [33] or P₃m̄l, Zhdanov symbol [2211]; $a = b = 3.643 \pm 0.002 \text{ \AA}$, $c = 17.683 \text{ \AA}$.

18R P₃m̄l, Zhdanov symbol [1212]₃; $a = b = 3.643 \text{ \AA}$, $c = 53.05 \text{ \AA}$.

The X-ray powder intensity data obtained from this sample are presented in Table III, where again a poor level of agreement exists between the experimentally observed and theoretically derived powder intensities. A number of X-ray reflections which were computed to possess high intensities, e.g. the 10.3 reflection of the 4H polytype, the 10.4 and the 10.5 of the 6H polytype, and the 10.7 and the 10.16 reflections from the 18R polytype, were in fact absent from the powder photograph. Additionally, the powder data are not in exact agreement with any mixture of the individual polytypes, nor of the 2H modification. It is clear that the structure of this sample has also

TABLE II A comparison of the experimentally observed powder data for SnS₂ crystal no. 2 (predominantly 4H polytype) with values calculated on the basis of the 4H structure (CuK α radiation)

Reflection <i>hkl</i>	θ_{obs} (deg)	θ_{calc} (deg)	d_{calc} (\AA)	I_{calc} (normalized)	I_{obs}
002	7.53	7.51	5.894	100.0	vvs
100	14.19	14.14	3.155	8.3	ms
101	14.69	14.65	3.048	61.3	m
004	—	15.16	2.947	3.3	abs
102	—	16.08	2.782	41.0	abs
103	18.11	18.26	2.460	74.0	vw
104	—	20.97	2.154	18.3	abs
006	—	23.10	1.965	5.7	abs
105	—	24.09	1.889	32.8	abs
110	25.05	25.04	1.822	31.5	m
112	26.27	26.29	1.740	23.2	mw
106	—	27.53	1.668	8.2	abs
200	—	29.25	1.577	1.1	abs
{ 201	29.60	29.54	1.564	8.5	vw
{ 114		29.84	1.549	3.6	
202	—	30.39	1.524	6.0	abs
107	—	31.26	1.486	7.2	abs
{ 008	31.71	31.54	1.474	5.1	vw
{ 203		31.78	1.464	14.0	
204	—	33.66	1.391	4.3	abs
{ 116	35.29	35.25	1.336	9.5	vw
{ 108		35.27	1.335	1.3	
205	36.04	36.01	1.311	9.6	vw
206	—	38.81	1.230	2.9	abs
109	—	39.59	1.210	3.6	abs
210	—	40.28	1.192	0.9	abs
211	—	40.52	1.186	6.7	abs
00.10	—	40.84	1.179	1.0	abs

TABLE III A comparison of the experimentally observed powder data for SnS₂ crystal sample no. 3 with values computed on the basis of the 4H, 6H and 18R polytypes (CuK α radiation)

θ_{obs} (deg)	[22]; 4H structure			6H structures				[1212] ₃ ; 18R structure			I_{obs}
	Reflection	θ_{calc} (deg)	I_{calc} (norm)	Reflection	θ_{calc} (deg)	I_{calc} (norm)		Reflection	θ_{calc} (deg)	I_{calc} (norm)	
						[2211]	[33]				
7.46	002	7.51	100.0	003	7.51	100.0	100.0	009	7.51	100.0	vs
14.24	100	14.14	8.3	100	14.14	11.1	0.0	101	14.17	5.7	ms
				101	14.37	12.3	1.8				
				10 $\bar{1}$	14.37	12.3	44.1				
				102	14.24	22.1					
—	101	14.65	61.3					104	14.55	4.8	abs
—								10 $\bar{5}$	14.77	12.0	abs
—				{ 102	15.04	15.0	28.7				abs
—				{ 10 $\bar{2}$	15.04	15.0	4.7				abs
—	004	15.16	3.3	006	15.16	3.3	3.3	00.18	15.16	3.3	abs
—								107	15.35	38.2	abs
—								10 $\bar{8}$	15.70	1.8	abs
16.19	102	16.09	41.0	{ 103	16.09	9.2	5.1				vw
—				{ 10 $\bar{3}$	16.09	40.3	5.1				
—								10.10	16.52	26.3	abs
—								10.1 $\bar{1}$	16.98	29.2	abs
—				{ 104	17.47	17.2	3.0				abs
—				{ 10 $\bar{4}$	17.47	17.2	62.2				abs
—	103	18.26	74.0					10.13	17.99	7.0	abs
—								10.1 $\bar{4}$	18.54	12.0	abs
—				{ 105	19.11	15.3	39.7				abs
—				{ 10 $\bar{5}$	19.11	15.3	0.7				abs
—								10.16	19.71	32.2	abs
—								10.17	20.33	2.1	abs
—	104	20.97	18.3	106	20.97	24.5	0.0				abs
—								10.19	21.63	1.8	abs
22.25								10.20	22.31	22.3	vw
—				{ 107	23.01	8.8	0.4				abs
—				{ 10 $\bar{7}$	23.01	8.8	23.1				abs
—	006	23.10	5.7	009	23.10	5.7	5.7	00.27	23.10	5.7	abs
—								10.22	23.73	5.8	abs
—	105	24.09	32.8					10.23	24.46	2.9	abs
24.84	110	25.04	31.5	110	25.04	31.5	31.5	110	25.04	31.5	s
—				{ 108	25.20	5.9	21.1				abs
—				{ 10 $\bar{8}$	25.20	5.9	0.9				
—								10.25	25.96	8.6	abs
26.26	112	26.29	23.2	113	26.29	23.2	23.2	119	26.29	23.2	m

been influenced by the powdering process. Intensities calculated on the assumption that polytype transformation had taken place failed to give improved correlation between observed and calculated data.

It is thought that the grinding process used to prepare X-ray powder specimens influences the

structures, except for the 2H modification, of polytypic crystals of SnS₂. The effect is not satisfactorily explained by polymorphic phase changes to other tin-sulphur compounds (such phase changes would in any case normally require the application of high temperatures). Nor is the effect explained fully by the formation of alternative

polytypes.

For the polytypic crystals examined, powder lines of notable intensity from higher polytypes are usually overlapped by strong 2H reflections, the only exception being the 1 0. 1 reflection from the 4H structure. This is probably why only one SnS₂ polytype other than the basic 2H modification has ever been reported from powder studies alone, this latter being the 4H polytype observed by Guenter and Oswald [3]. Ishizawa and Fujiki [4–6] have also examined polytypes of SnS₂ by powder and Weissenberg techniques and infer in their work that single crystal X-ray examination is to be preferred if reliable structure analysis is to be achieved. Terhell and Lieth [7] have reported structural alterations similar to those described above in their studies of GaSe layer crystals.

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Optical absorption of transition metals in alkali lime germanosilicate glasses

The absorption loss in alkali lime germanosilicate optical fibres, made by the double crucible method, is due in general to the presence in the glass of OH groups and transition metals, in concentrations in the p.p.m. and p.p.b. range respectively [1, 2]. In the course of an investigation aimed at lowering the loss of these optical fibres, the absorption spectra of the different transition metal ions in alkali lime germanosilicate glass were measured.

Typical core glasses of molar composition 0.44 SiO₂ 0.29 GeO₂ 0.02 Al₂O₃ 0.10 CaO 0.14 Na₂O 0.1 K₂O with 0.05% As₂O₃ acting as a fining agent were melted in Pt–10% Rh crucibles in a resistance furnace. Very pure starting

materials* with transition metal impurities below 0.01 p.p.m. were used. Transition metal compounds were added to the batch to give a concentration of 20 p.p.m. (V, Cr, Co, Ni, and Cu) or 200 p.p.m. (Ti, Mn and Fe). The melt was bubbled for one hour at 1400°C after which the melt was allowed to stand for one hour at 1300°C for fining. The bubble free melt was poured into graphite moulds and annealed at 620°C for one hour. This melting procedure was chosen because it had previously proved possible to minimize the absorption loss of optical fibres by using pure glasses prepared in this manner in fused silica crucibles [2]. The absorption spectra were recorded from 350 to 1500 nm with a Cary 17 spectrophotometer on 5 to 10 cm polished rods. This range incorporates the wavelengths at which it is proposed to use optical fibres (800 to 1300 nm)

*SiO₂, CaCO₃, Na₂CO₃, K₂CO₃ and As₂O₃ (Merck, FO Optipur), GeO₂ (Hoboken-Belgium extra pure) and Al₂O₃ (J. T. Baker, Ultrex).